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purpose of the lab; what question(s) are you trying to answer?

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*include any preliminary observations or background information about the subject.

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Answer (1 of 5): Molar mass of Chlorine(Cl) is 35.45 g/mol, so Cl₂

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has an atomic mass of 70.9 g/mol, now with simple cross dividing we can get $70.9/x=0.5273/1$, we can ...

A metal (M) forms a compound with the formula MCl₂. If the ...

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administration, minimal levels of iodoform may be found in disinfectants and it is more primarily used for veterinary purposes.

Iodoform | CHI₃ - PubChem

Methylene blue, also known as methylthioninium chloride, is a salt used as a medication and dye. As a medication, it is mainly used to treat methemoglobinemia. Specifically, it is used to treat methemoglobin levels that are greater than 30% or in which there are symptoms despite oxygen therapy. It has previously been used for cyanide poisoning and urinary tract infections, but this use is no ...

Methylene blue - Wikipedia

In equation , $[M]$ e is the equilibrium monomer concentration in moles per litre, T is the reaction temperature in Kelvin (K), ΔH is the enthalpy of reaction in kilocalories per mole, ΔS is the ...

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Olefin metathesis-based chemically recyclable polymers

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Balancing chemical equations practice questions and answers

CRC Handbook of Chemistry and Physics. 79th ed. Boca Raton, FL: CRC Press Inc., 1998-1999., p. 6-170 Hazardous Substances Data Bank (HSDB) 0.39 mm^2/s at 25 °C

Dibromomethane | CH_2Br_2 - PubChem

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Acceleration - Physics Classroom

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