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121 Section 8.1 The

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240–247 Practice

Problems page 244

Draw the Lewis

structure for each

molecule. 1. PH₃, H₂, H₂

H—H, H₂, P₄ respectively,

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for single, double, and

triple P — — 2. H 2 S H

H H — H S S — 3. HCl H

Cl H — Cl 4. CCl 4 Cl Cl

Cl C C — —

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determine the mass of one substance involved in a chemical reaction if given the mass of any one of the other substances involved in the reaction. in other words . . .

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SOLUTIONS MANUAL

Section 11.1

Assessment page 372

5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are

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related. The coefficients in the balanced equation indicate the molar relationship between each ...

Stoichiometry

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